

EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS) pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

# **COURSE DESCRIPTION CARD - SYLLABUS**

Course name		
Composites		
Course		
Field of study		Year/Semester
Chemical and process en	2/3	
Area of study (specializat	Profile of study	
Bioprocesses and biomat	general academic	
Level of study		Course offered in
Second-cycle studies		Polish
Form of study		Requirements
full-time		compulsory
Number of hours		
Lecture	Laboratory classes	Other (e.g. online)
30	15	
Tutorials	Projects/seminars	
Number of credit points		
6		
Lecturers		
Responsible for the course/lecturer: Res		esponsible for the course/lecturer:
Mariola Sądej, PhD, MSc Jak		akub Zdarta, PhD, MSc

#### **Prerequisites**

Knowledge of basic chemistry, organic chemistry and inorganic chemistry as well as knowledge and skills in the field of chemical technology and polymer technology.

### **Course objective**

To get a theoretical and practical knowledge about the production and properties of polymeric and inorganic composites, as well as use of organic and inorganic fillers. Understanding of the basic industrial processes and operations related to the prudction technology and the characteristics of composite materials. Ability to select raw materials and precursors for synthesis of a desired systems. Strengthening knowledge through practical exercises.

#### **Course-related learning outcomes**

#### Knowledge

K\_W04. The student has knowledge about complex chemical processes, including the appropriate selection of materials, raw materials, apparatus and equipment for the implementation of chemical processes and characterization of the products obtained.



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K\_W07. The student has knowledge of the novel chemical and material technologies, including technologies of advanced materials and nanomaterials, knows the current trends in the development of chemical industrial processes

K\_W08. The student knows modern methods of testing the structure and properties of materials, necessary to characterize raw materials and products of the chemical and related industries.

Skills

K\_U02. The student has the ability to work in a team and lead a team.

K\_U06. The student has the ability to present research results in the form of a report, dissertation or presentation.

K\_U11. The student has the ability to adapt knowledge in chemistry and related fields to solve technological problems and to plan new industrial processes, not only chemical

K\_U18. The student is able to critically evaluate the results of experimental research and determine the direction of further research leading to solving problems in the field of chemical engineering, process equipment and industrial technologies.

### Social competences

K\_K01. The student understands the need for lifelong learning; can inspire and organize the learning process of others; is aware of the importance and non-technical aspects and effects of engineering activities, including its impact on the environment, and the associated responsibility for the decisions taken.

K\_K03. The student is able to interact and work in a group, taking on different roles.

K\_K07. The student is aware of the social role of a technical university graduate, and in particular understands the need to formulate and convey to the public, in particular through the mass media, information and opinions on the achievements of technology and other aspects of engineering activities; endeavors to provide such information in a manner that is universally understandable and gives reasons for different points of view.

### Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Final exam related to the content of the lecture. Reports after laboratory exercises.

### **Programme content**

Lectures include:

Inorganic matrix composites: general information on inorganic composite materials; review of methods of obtaining of inorganic composite systems; surface functionalization of composite oxide materials; physicochemical, dispersion and morphological characteristics of composite oxide systems and their derivatives; oxide composites with defined properties for use in various processes; directions of use of advanced powder substances.



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Polymer matrix composites: basic information on polymer composites - definition and components as well as used precursors; methods of strengthening polymers; preparation and types of composites and their characteristics; methods of synthesis of polymer composites; nanocomposites; differences in the structure and properties of composites and nanocomposites; physical, chemical and mechanical properties of (nano) composites, their processing and recycling; application of (nano) polymer composites with particular emphasis on composites in medicine and dentistry; basic information about development trends in the field of synthesis of composite materials.

Laboratory exercises include:

Inorganic matrix composites: preparation of composite oxide materials, physicochemical and dispersion characteristics of composite oxide systems and their derivatives, methods of surface functionalization of hybrid oxide materials, colorimetric characteristization of pigment systems, determination of sorption properties of oxide systems.

Polymer-based composites: curing dental composites, obtaining and testing the physicochemical and mechanical properties of composite materials used in medicine; identification of composite materials and fillers used.

### **Teaching methods**

Lectures, laboratory excercises.

### Bibliography

#### Basic

1. A. Boczkowska, J. Kapuściński, Z. Lindemann, D. Witemberg-Perzyk, S. Wojciechowski, Kompozyty, Oficyna Wydawnicza Politechniki Warszawskiej, 2003.

2. G. Wypych, Handbook of fillers, ChemTec Publishing, 2010.

3. G. Wilde, Nanostructured Materials, Elsevier, 2009.

4. E.F. Vansant, P. Van Der Voort, K.C. Vrancken, Characterization and Chemical Modification of the Silica Surface, Elsevier, 1997.

### Additional

- 1. Research artciles related to the topic of course.
- 2. A. Jess, Chemical Technology: An Integral Textbook, Wiley VCH, 2012.
- 3. J.A. Moulijn, Chemical Process Technology, Wiley VCH, 2013.



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## Breakdown of average student's workload

	Hours	ECTS
Total workload	150	6,0
Classes requiring direct contact with the teacher	90	4,0
Student's own work (literature studies, preparation for laboratory	60	2,0
classes/tutorials, preparation for tests/exam) <sup>1</sup>		

<sup>&</sup>lt;sup>1</sup> delete or add other activities as appropriate